

Equations Table:

$N_A = 6.02 \times 10^{23} \text{ mol}^{-1}$		$PV = nRT$
$R = 8.314 \text{ J/mol} \cdot \text{K}$		$PV = NkT$
$k = 1.38 \times 10^{-23} \text{ J/K}$	$\Delta L = \alpha L_0 \Delta T$	$T(\text{K}) = T(^{\circ}\text{C}) + 273.15$
	$\Delta V = \beta V_0 \Delta T$	$1 \text{ cal} = 4.186 \text{ J}$
$\sigma = 5.67 \times 10^{-8} \text{ W/m}^2 \cdot \text{K}^4$		$P = P_0 + \rho gh$

$Q = mc\Delta T$	$\Delta U = Q - W$	$\frac{dQ}{dt} = kA \left \frac{dT}{dx} \right $
$Q = mL$	$dU = dQ - dW$	
$dW = PdV; W = \int_i^f PdV$	$W = nRT \ln\left(\frac{V_f}{V_i}\right)$	$\frac{\Delta Q}{\Delta t} = \sigma AeT^4$

$P = \frac{2}{3} \frac{N}{V} \left(\frac{1}{2} m \langle v^2 \rangle_{av} \right)$	$dU = nC_V dT$	
$\frac{1}{2} m \langle v^2 \rangle_{av} = \frac{3}{2} kT$	$C_p - C_V = R$	$f(v) = 4\pi \left(\frac{m}{2\pi kT} \right)^{3/2} v^2 e^{-mv^2/2kT}$
	$\gamma = C_p/C_V$	
$U = \frac{3}{2} NkT = \frac{3}{2} nRT$	$PV^\gamma = \text{constant}$	

$W = Q_h - Q_c $	$dS = \frac{dQ_r}{T}$	
$e = \frac{W}{ Q_h }$	$\Delta S = \int_i^f \frac{dQ_r}{T}$	$\Delta S_{total} = \Delta S_{system} + \Delta S_{environment} \geq 0$
$e_c = 1 - \frac{T_c}{T_h}$	$\Delta S = nC_V \ln(T_f/T_i) + nR \ln(V_f/V_i)$	