

Wave Equation for a Quantum Free Particle

Thus, a correct quantum wave function for a quantum free particle must satisfy this quantum dispersion relation for k and ω :

$$\hbar\omega = \frac{\hbar^2 k^2}{2m} \quad (*) \quad (\text{non-relativistic})$$

We now assume the same fundamental sinusoidal form for the wave function of a quantum free particle with mass m , momentum $p = \hbar k$ and energy $E = \hbar\omega$:

$$\Psi(x, t) = A \cos(kx - \omega t) + B \sin(kx - \omega t)$$

Recall from our discussion on the mechanical wave, we have the following:

$\frac{\partial}{\partial x} \rightarrow$	take out an overall k factor from $\Psi(x, t)$	$\frac{\partial}{\partial t} \rightarrow$	take out an overall $-\omega$ factor from $\Psi(x, t)$
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Wave Equation for a Quantum Free Particle

So, from the quantum dispersion relation,

$$\frac{\hbar^2 k^2}{2m} = \hbar \omega$$

We can argue that the PDE for the quantum wave function for this free particle must involve:

$$\frac{\partial^2}{\partial x^2} \quad \quad \quad -\frac{\partial}{\partial t}$$

Putting in other constants so that units are consistent and one additional dimensionless “fitting” constant C , we then have this trial wave equation,

$$\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} = -C\hbar \frac{\partial \Psi(x,t)}{\partial t}$$

Wave Equation for a Quantum Free Particle

Now, we substitute our trial quantum wave function

$$\Psi(x, t) = A \cos(kx - \omega t) + B \sin(kx - \omega t)$$

into the proposed wave equation to solve for the “fitting” constant C :

$$\begin{aligned} \frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x, t)}{\partial x^2} &= \frac{\hbar^2}{2m} \left[-Ak^2 \cos(kx - \omega t) - Bk^2 \sin(kx - \omega t) \right] \\ &= \frac{\hbar^2 k^2}{2m} \left[-A \cos(kx - \omega t) - B \sin(kx - \omega t) \right] \\ -C\hbar \frac{\partial \Psi(x, t)}{\partial t} &= -C\hbar \left[-A(-\omega) \sin(kx - \omega t) + B(-\omega) \cos(kx - \omega t) \right] \\ &= \hbar\omega \left[CB \cos(kx - \omega t) - CA \sin(kx - \omega t) \right] \end{aligned}$$

Wave Equation for a Quantum Free Particle

Equating the two terms and using the equality $\frac{\hbar^2 k^2}{2m} = \hbar\omega$, we have,

$$\frac{\hbar^2 k^2}{2m} [-A \cos(kx - \omega t) - B \sin(kx - \omega t)] = \hbar\omega [CB \cos(kx - \omega t) - CA \sin(kx - \omega t)]$$

In order for this equality to be true for all (x, t) , all coeff's for cos and sin must equal to each other,

$$\longrightarrow \begin{cases} -A = CB \\ B = CA \end{cases}$$

Substituting the first eq into the second, we have,

$$B = C(-CB) \rightarrow C^2 = -1$$

Thus, the “fitting” constant is $C = i$ where $i = \sqrt{-1}$.

Wave Equation for a Quantum Free Particle

Then, finally, putting everything together, we have the desired wave equation for a quantum free particle,

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} = i\hbar \frac{\partial \Psi(x,t)}{\partial t}$$

This is the 1D **Schrodinger's Equation** for a free particle.

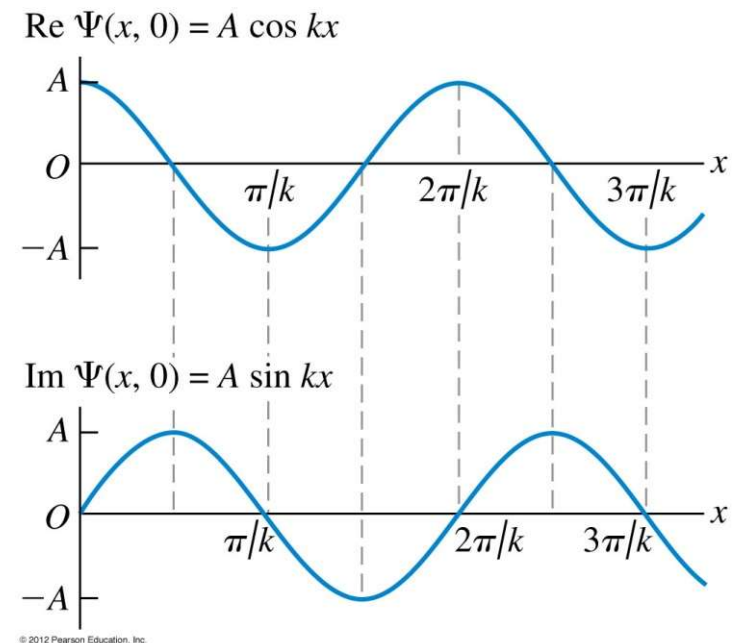
With $B = CA = iA$, the free particle quantum wave function can also be written in a compact exponential form using the Euler's formula,

$$\Psi(x,t) = A \left[\cos(kx - \omega t) + i \sin(kx - \omega t) \right]$$

$$\Psi(x,t) = A e^{i(kx - \omega t)} \quad (\text{quantum wave function for a free particle})$$

Free Particle Wave Function & Uncertainty Principle

So, the wave function for a free particle is a *complex* function with *sinusoidal* real and imaginary parts



A quantum *free* particle in principle exists in *all* space $[-\infty, +\infty]$,

➔ $\Delta x = \infty$ & $\Delta t = \infty$ (wave function extends into all space & time)
but $\Delta p = 0$ & $\Delta E = 0$ (energy and momentum is fixed)

Note: $\Delta x \Delta p \geq \hbar/2$ & $\Delta t \Delta E \geq \hbar/2$ can still be satisfied.

More Realistic Particle (Wave Packets)

Under more realistic circumstance, a particle will have a relatively well defined position and momentum so that *both* Δx and Δp will be *finite* with limited spatial extents.

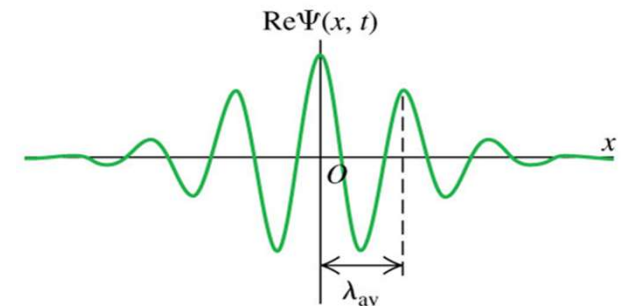
A more *localized* quantum particle can *not* be a pure sine wave and it must be described by a **wave packet** with a combination of many sine waves.

$$\Psi(x, t) = \int_{-\infty}^{\infty} A(k) e^{i(kx - \omega t)} dk$$

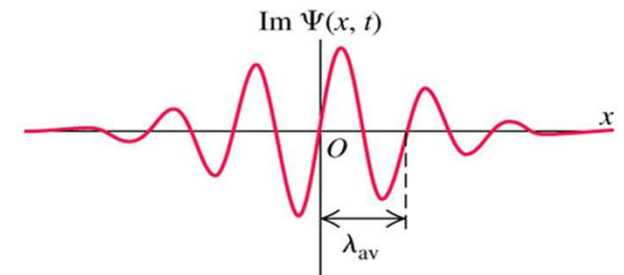
(a linear combination of many sine waves.)

The coefficient $A(k)$ gives the relative proportion of the various sine waves with diff. k (wave number).

(a) Real part of the wave function at time t

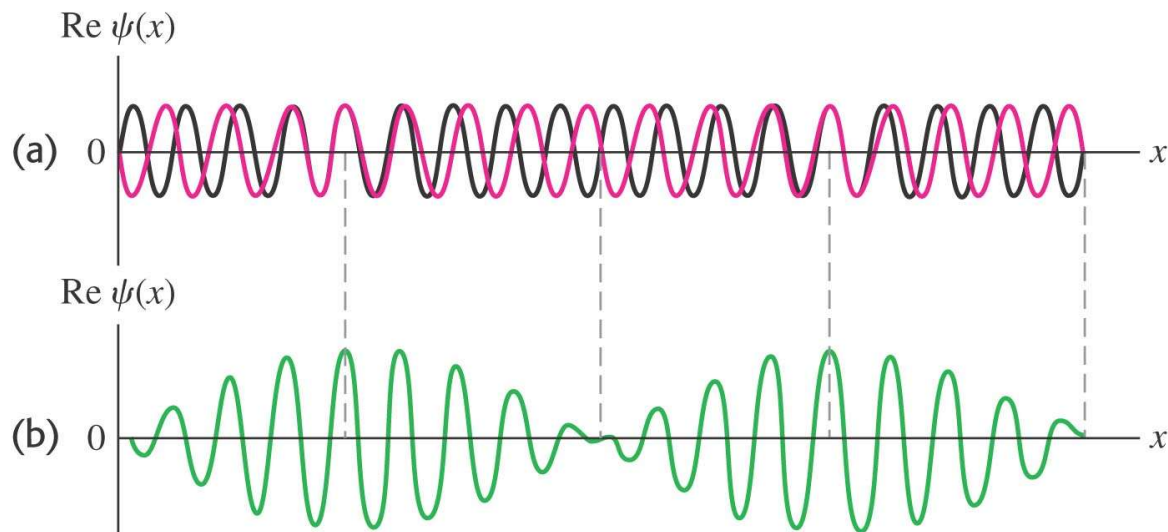


(b) Imaginary part of the wave function at time t



Wave Packets

Recall: Combination of *two* sine waves
→ more localized than a pure sine wave.



Wave Packets

A **wave packet** is a linear combination of many sine waves.

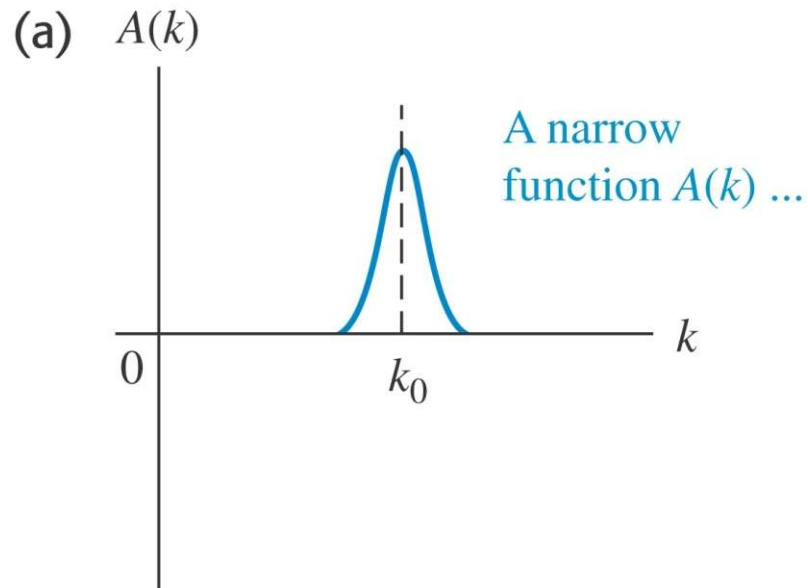
$$\Psi(x, t) = \int_{-\infty}^{\infty} A(k) e^{i(kx - \omega t)} dk$$

Recall
$$P = \frac{h}{\lambda} = \frac{h}{2\pi} \frac{2\pi}{\lambda} = \hbar k$$

So, the wave number k is directly proportional to the momentum of the quantum particle.

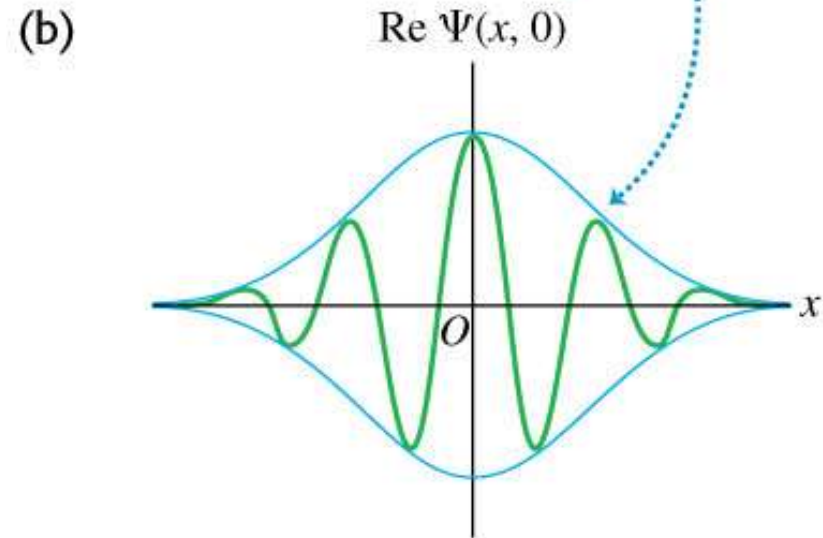
Wave Packets (characteristic)

$A(k)$ gives the spread
in k (momentum P)



Δp smaller

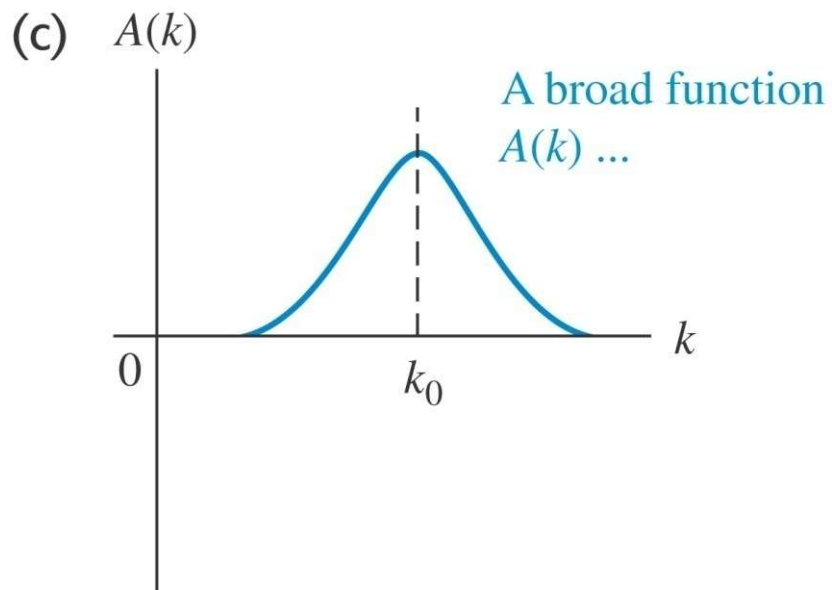
... gives a wave function $\Psi(x, 0)$
with a broad spatial extent.



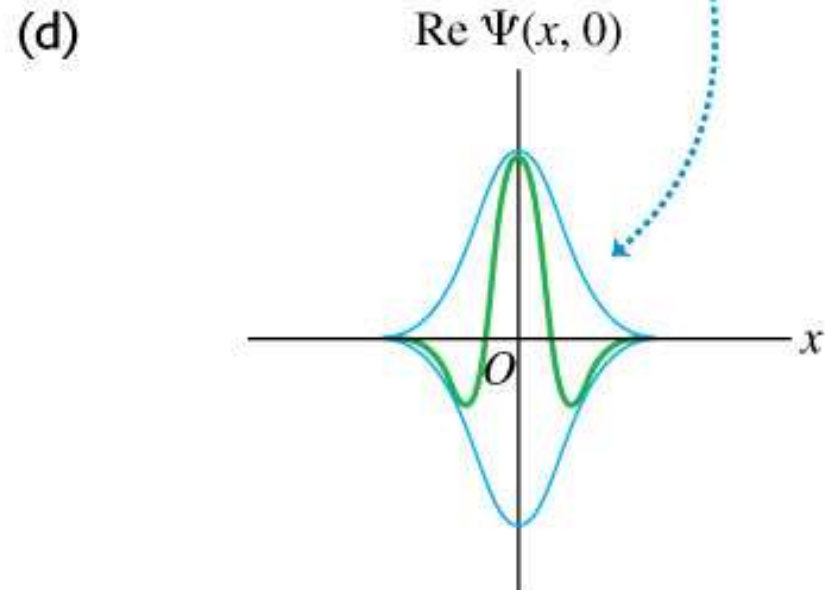
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Δx bigger

Wave Packets (characteristic)



Δp bigger



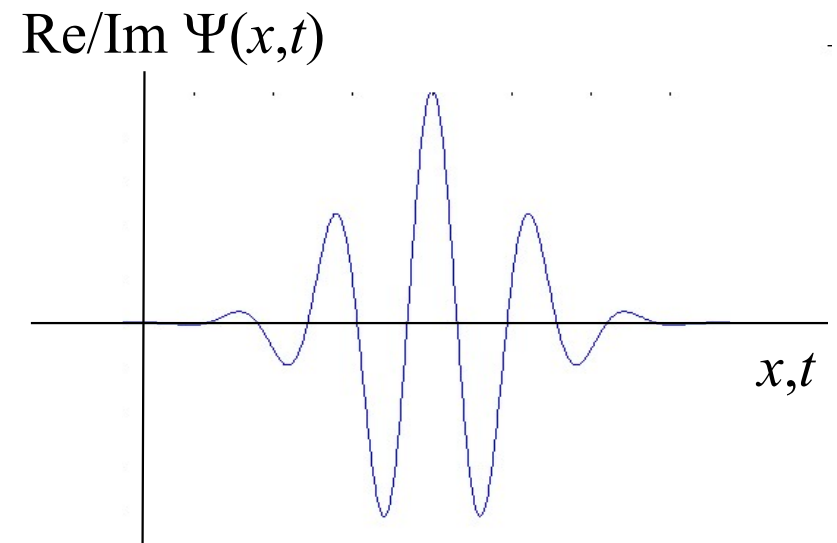
Δx smaller

The is consistent with: $\Delta x \Delta p \geq \hbar$!

Quantum Wave Function

In QM, the **matter wave** postulated by de Broglie is described by a *complex-valued wavefunction* $\Psi(x,t)$ which is the fundamental descriptor for a quantum particle.

1. Its absolute value squared $|\Psi(x,t)|^2 dx$ gives the probability of finding the particle in an infinitesimal volume dx at time t .
2. For any Q problem:
 - The goal is to find $\Psi(x,t)$ for the particle for all time.
 - Physical interactions involves “operations” (O) on this wave function: $O \Psi(x,t)$
 - Experimental measurements will involve the “products”, $\Psi^*(x,t) O \Psi(x,t)$



$\Psi(x,t)$ is a complex-valued function of space and time.

The **General** 1D Schrodinger Equation for a Quantum Particle (not free)

Recall that the wave equation for a quantum *free* particle is

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} = i\hbar \frac{\partial \Psi(x,t)}{\partial t}$$

$$\text{KE} = \text{Total E}$$

It is basically a statement on the conservation of energy!

The **General** 1D Schrodinger Equation for a Quantum Particle (not free)

Now, for the general case:

The **Schrodinger equation** is again a statement on the conservation of energy.

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + U(x) \Psi(x,t) = i\hbar \frac{\partial \Psi(x,t)}{\partial t}$$

$$\text{KE} \quad + \quad \text{PE} \quad = \text{Total E}$$

As we have seen for the free quantum particle case,

- the *first term* (2nd order spatial derivative term) in the Schrodinger equation is associated with the Kinetic Energy of the particle
- the last term (the 1st order time derivative term) is associated with the total energy of the particle
- now we also include the Potential Energy term $U(x) \Psi(x)$

The Schrodinger Equation

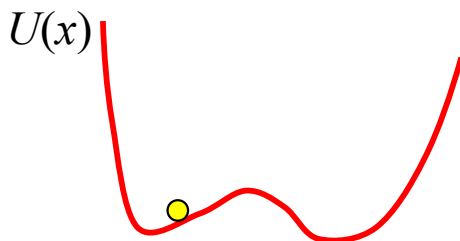
In Classical Mechanics, we have the Newton's equation which describes the trajectory $\mathbf{x}(t)$ of a particle:

$$\mathbf{F} = m\ddot{\mathbf{x}}$$

In EM, we have the wave equation for the propagation of the E , B fields:

$$\frac{\partial^2 E, B}{\partial x^2} = \frac{1}{c^2} \frac{\partial^2 E, B}{\partial t^2} \quad (\text{derived from Maxwell's eqns})$$

In QM, **Schrodinger equation** prescribes the evolution of the wavefunction for a particle in time t and space x under the influence of a potential energy $U(x)$,



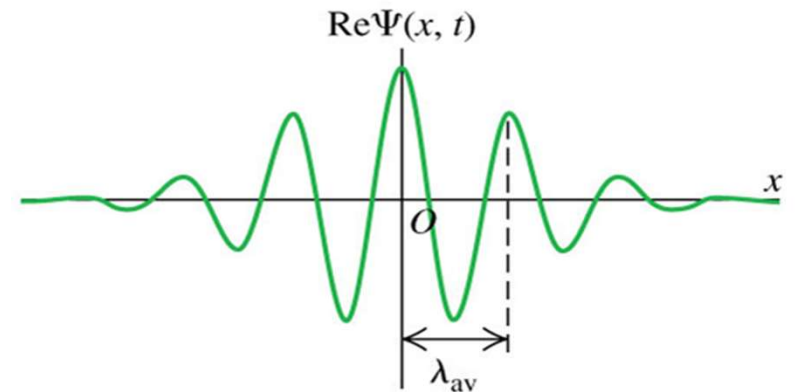
$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + U(x)\Psi(x,t) = i\hbar \frac{\partial \Psi(x,t)}{\partial t}$$

(general 1D Schrödinger equation)

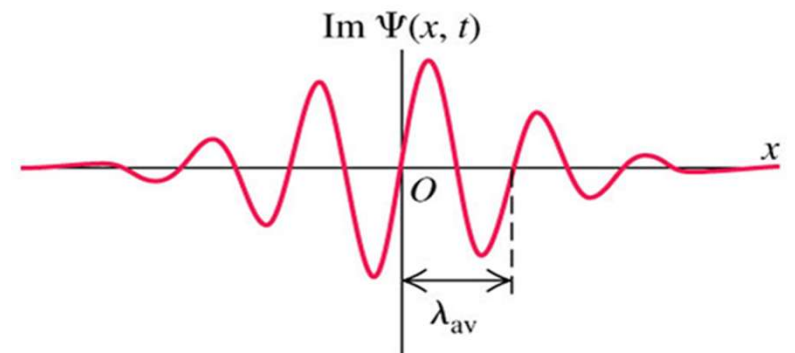
Quantum Wave Function

The **wave function** $\Psi(x, t)$ for a quantum particle is a *complex* function with *sinusoidal* real and imaginary parts

(a) Real part of the wave function at time t



(b) Imaginary part of the wave function at time t

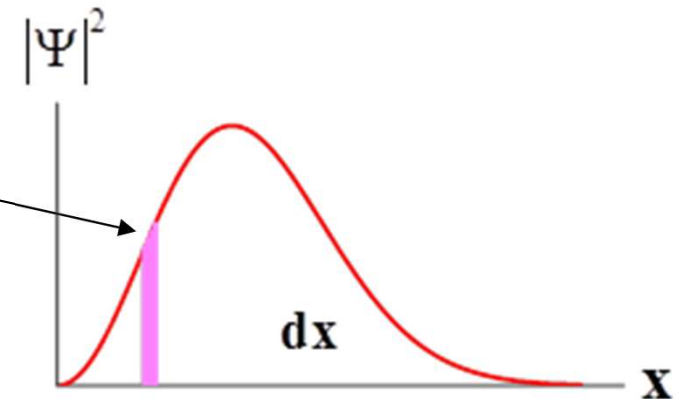


Wave Function and Probability

$$|\Psi(x,t)|^2 = \Psi^*(x,t)\Psi(x,t)$$

is the **probability distribution function** for the quantum particle.

In other words, $|\Psi(x,t)|^2 dx$ (shaded area) is the *probability* in finding the particle in the interval $[x, x + dx]$ at time t .



Since $p(x)dx \equiv |\Psi(x,t)|^2 dx$ is a probability, it has to be **normalized** !

$$\longrightarrow \int_{-\infty}^{+\infty} p(x)dx = \int_{-\infty}^{+\infty} |\Psi(x,t)|^2 dx = 1$$

(At any instance of time t , the particle must be somewhere in space !)

Stationary States

For most problems, we can factor out the *time dependence* by assuming the following harmonic form for the time dependence,

$$\Psi(x, t) = \psi(x)e^{-i\omega t}$$

[Recall the free particle case: $\Psi(x, t) = Ae^{i(kx - \omega t)} = Ae^{ikx}e^{-i\omega t}$.]

With $\omega = E / \hbar$, we can rewrite the time exponent in terms of E ,

$$\Psi(x, t) = \psi(x)e^{-iEt/\hbar}$$

$\Psi(x, t)$ is a state with a definite energy E and is called a **stationary state**.

$\psi(x)$ is called the *time-independent* wave function.

The Time-Independent Schrödinger Equation

Substituting this factorization into the general time-dependent Schrödinger Eq, we have

$$\text{RHS} \longrightarrow i\hbar \frac{\partial \Psi(x,t)}{\partial t} = i\hbar \frac{\partial}{\partial t} \left(\psi(x) e^{-iEt/\hbar} \right) = i\hbar \psi(x) \left(\frac{-iE}{\hbar} \right) e^{-iEt/\hbar} = E\psi(x) e^{-iEt/\hbar}$$

and,

$$\text{LHS} \longrightarrow \frac{\partial^2 \Psi(x,t)}{\partial x^2} = \frac{d^2 \psi(x)}{dx^2} e^{-iEt/\hbar}$$

$$-\frac{\hbar^2}{2m} \frac{d^2 \psi(x)}{dx^2} e^{-iEt/\hbar} + U(x) \psi(x) e^{-iEt/\hbar} = E\psi(x) e^{-iEt/\hbar}$$

$$-\frac{\hbar^2}{2m} \frac{d^2 \psi(x)}{dx^2} + U(x) \psi(x) = E\psi(x)$$

(time dependence can be cancelled out !)

(time-independent Schrödinger equation)

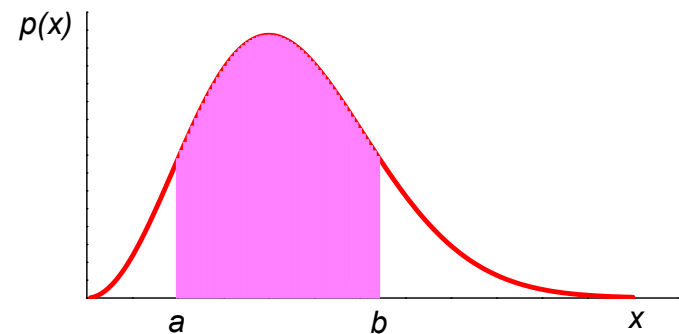
More on (time-independent) Wavefunction

Note that, $|\Psi(x,t)|^2 = \Psi^*(x,t)\Psi(x,t) = \psi^*(x)e^{-iEt/\hbar}\psi(x)e^{iEt/\hbar}$
 $= \psi^*(x)\psi(x)e^{i(-Et/\hbar+Et/\hbar)} = \psi^*(x)\psi(x) = |\psi(x)|^2$

So, in general, the probability in finding the particle in the interval $[a,b]$ is given by:

$$P_{ab} = \int_a^b |\psi(x)|^2 dx$$

Note: $\psi(x)$ is *not* the probability density
 $|\psi(x)|^2$ is the probability density.



More on (time-independent) Wavefunction

Other physical **observables** can be obtained from $\psi(x)$ by the following operation:

example (position x): $\langle x \rangle = \int_{-\infty}^{\infty} xp(x)dx = \int_{-\infty}^{\infty} x|\psi(x)|^2 dx$

- $\langle x \rangle$ is called the **expectation value** (of x): it is the experimental value that one should expect to measure in real experiments !

In general, any experimental observable (position, momentum, energy, etc.) $O(x)$ will have an expectation value given by:

$$\langle O \rangle = \int_{-\infty}^{\infty} O(x)|\psi(x)|^2 dx \quad O \text{ can be } x, p, E, \text{ etc.}$$

- Note:
- Expectation values of physically measurable functions are the only experimentally accessible quantities in QM.
 - Wavefunction $\psi(x)$ itself is not a physically measurable quantity.

Solving QM Problems with (time-independent) Schrodinger Equation

Given: A particle is moving under the influence of a potential $U(x)$.

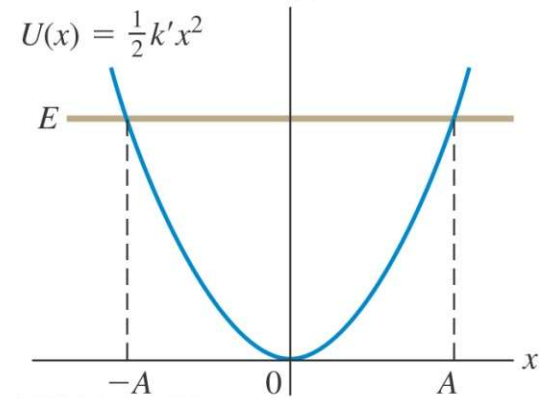
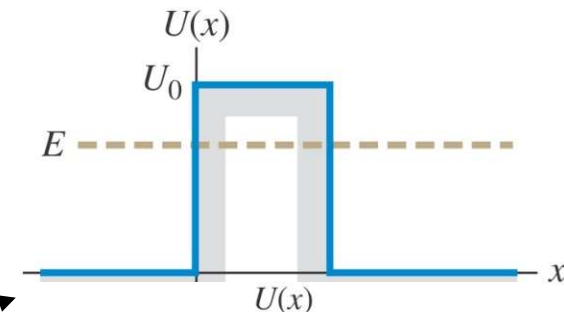
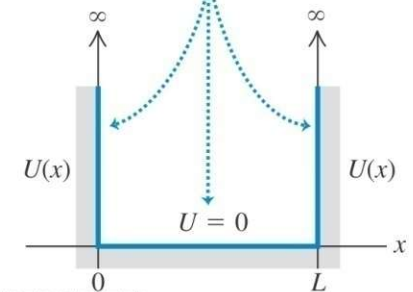
Examples: • Free particle: $U(x) = 0$

• Particle in a box: $U(x) = \begin{cases} 0, & 0 \leq x \leq L \\ \infty, & \text{elsewhere} \end{cases}$

• Barrier: $U(x) = \begin{cases} U_0, & 0 \leq x \leq L \\ 0, & \text{elsewhere} \end{cases}$

• HMO: $U(x) = \frac{1}{2}k'x^2$

The potential energy U is zero in the interval $0 < x < L$ and is infinite everywhere outside this interval.



Solving QM Problems with (time-independent) Schrodinger Equation

The general process in solving a quantum problem under the influence of a given potential $U(x)$ involves:

Solve time-independent Schrodinger equation for $\psi(x)$ as a function of energy E , with the restrictions:

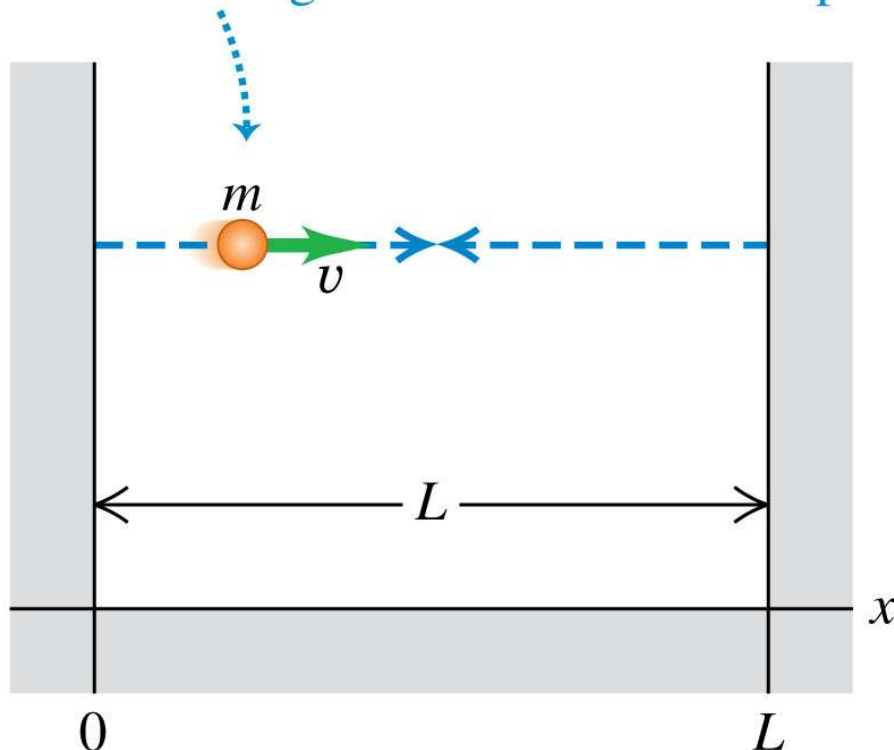
- $\psi(x)$ and $\frac{d\psi(x)}{dx}$ are *continuous* everywhere for smooth $U(x)$.
- $\psi(x)$ is *normalized*, i.e., $\int_{-\infty}^{\infty} |\psi(x)|^2 dx = 1$
- *Bounded* solution: $\psi(x) \rightarrow 0$ as $x \rightarrow \pm\infty$

Then, *expectation values* of physical measurable quantities can be calculated from $\psi(x)$.

Particle in a Box

Classical Picture

A particle with mass m moves along a straight line at constant speed, bouncing between two rigid walls a distance L apart.



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A 1- D box with *hard* walls:

$$U(0) = U(L) = \infty \quad (\text{non-penetrable})$$

A free particle *inside* the box:

$$U(x) = 0 \quad (\text{inside box})$$

No forces acting on the particle except at hard walls.

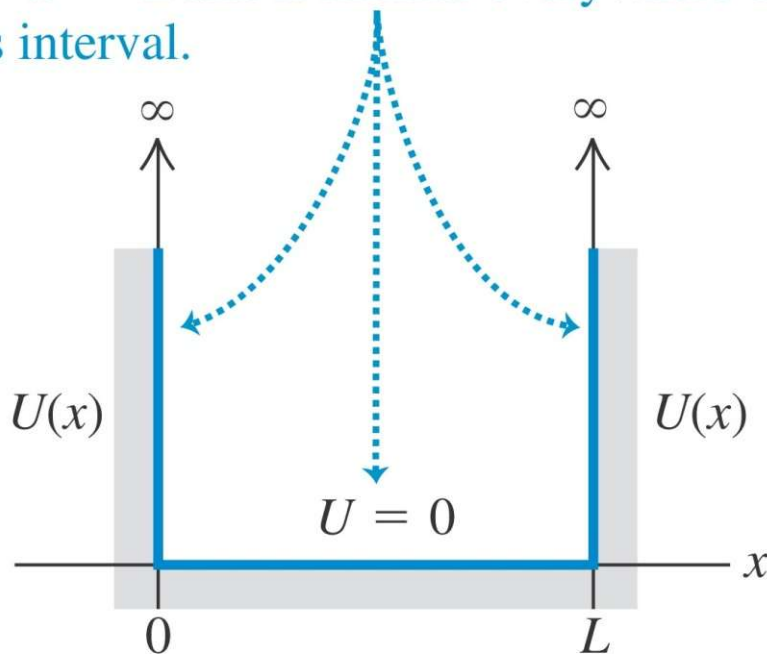
\mathbf{P} (in x) is conserved between bounces

→ $|\mathbf{P}|$ is fixed but \mathbf{P} switches sign between bounces.

Particle in a Box (Quantum Picture)

The situation can be described by the following potential energy $U(x)$:

The potential energy U is zero in the interval $0 < x < L$ and is infinite everywhere outside this interval.



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$$U(x) = \begin{cases} 0, & 0 < x < L \\ \infty, & \text{elsewhere} \end{cases}$$

The *time-independent* Schrodinger equation is:

$$-\frac{\hbar^2}{2m} \frac{d^2\psi(x)}{dx^2} + U(x)\psi(x) = E\psi(x)$$

Recall, this is basically

$$\text{KE} + \text{PE} = \text{Total E}$$

Problem statement: For this $U(x)$, what are the possible wave functions $\psi(x)$ and their corresponding allowed energies E ?

Wave functions for a Particle in a Box

Inside the box, $0 < x < L$, $U(x) = 0$, and the particle is free. From before, we know that the wave function for a free particle has the following form:

$$\psi_{inside}(x) = A_1 e^{ikx} + A_2 e^{-ikx} \quad (\text{linear combination of the two possible solutions.})$$

where A_1 and A_2 are constants that will be determined later. $\left(E = \frac{p^2}{2m} = \frac{\hbar^2 k^2}{2m} \right)$

Outside the box, $U(x) = \infty$, and the particle cannot exist outside the box and

$$\psi_{outside}(x) = 0 \quad (\text{outside the box})$$

At the boundary, $x = 0$ and $x = L$, the wavefunction has to be *continuous*:



$$\psi_{inside}(0, L) = \psi_{outside}(0, L) = 0$$

Wavefunctions for a Particle in a Box

Let see how this **boundary condition** imposes restrictions on the two constants, A_1 and A_2 , for the wave function.

Using the Euler's formula, we can rewrite the *interior* wave function in terms of sine and cosine:

$$\begin{aligned}\psi_{inside}(x) &= A_1 \overset{e^{+ikx}}{(\cos kx + i \sin kx)} + A_2 \overset{e^{-ikx}}{(\cos kx - i \sin kx)} \\ &= (A_1 + A_2) \cos kx + i(A_1 - A_2) \sin kx\end{aligned}$$

Imposing the **boundary condition** at $x = 0$,

$$\psi_{inside}(0) = (A_1 + A_2) \cos 0 + i \cancel{(A_1 - A_2)} \sin 0 = (A_1 + A_2) = 0$$



$$A_1 = -A_2$$



$$\psi_{inside}(x) = 2iA_1 \sin kx = C \sin kx \quad (\text{where } C=2iA_1)$$

Wavefunctions for a Particle in a Box

Now, consider the **boundary condition** at $x = L$:

$$\psi_{inside}(L) = C \sin kL = 0$$

For a non-trivial solution ($C \neq 0$), only certain sine waves with a particular choice of wave numbers (k) can satisfy this condition:

$$k_n L = n\pi \quad \text{or} \quad k_n = \frac{n\pi}{L}, \quad n = 1, 2, 3, \dots$$

This implies that the wavelengths within the box is *quantized* !

$$\lambda_n = \frac{2\pi}{k_n} = 2 \cancel{\pi} \frac{L}{n \cancel{\pi}} = \frac{2L}{n}, \quad n = 1, 2, 3, \dots$$

Allowed wavefunctions must have wavelengths exactly fit within the box !

Wavefunction for a Particle in a Box

Rewriting this, we have,

$$L = n \lambda_n / 2, \quad n = 1, 2, 3, \dots$$

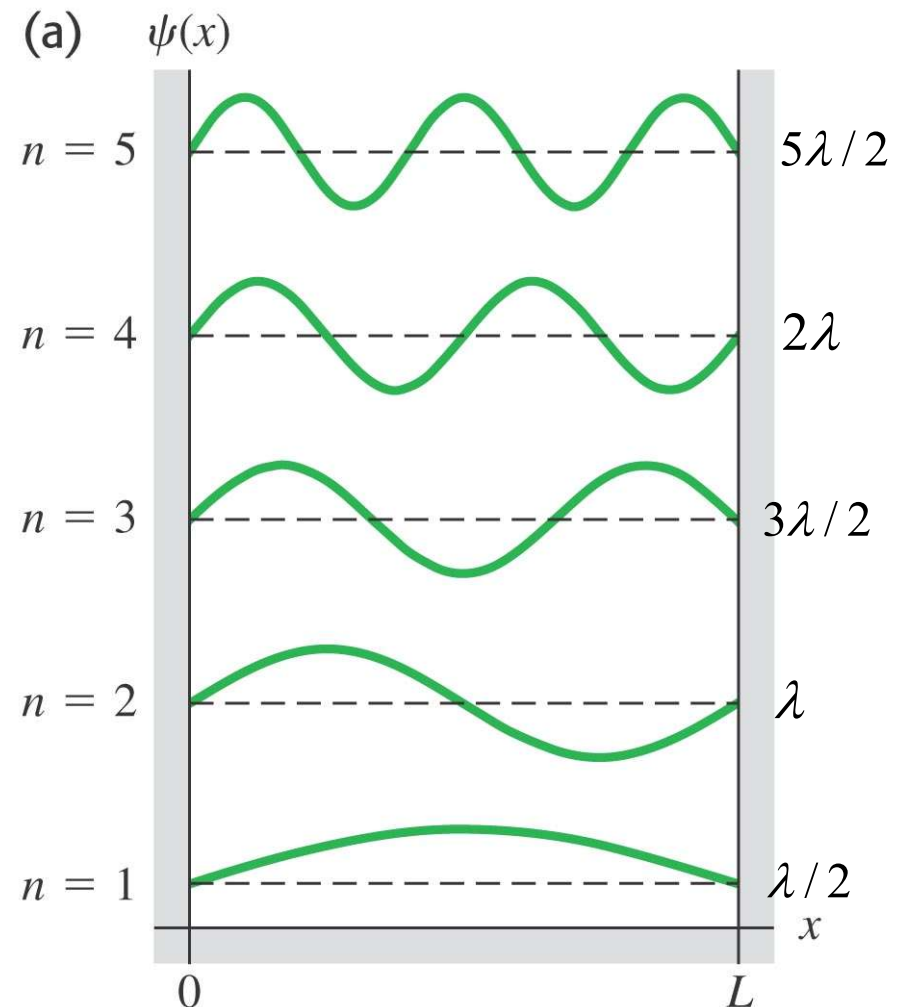
Graphically, it looks like



Since k_n is *quantized*, only a discrete set of $\psi_n(x)$ is allowed as solutions,

$$\psi_n(x) = C \sin(k_n x) = C \sin\left(\frac{n\pi}{L} x\right),$$
$$n = 1, 2, 3, \dots$$

(similar to *standing waves* on a cramped string)

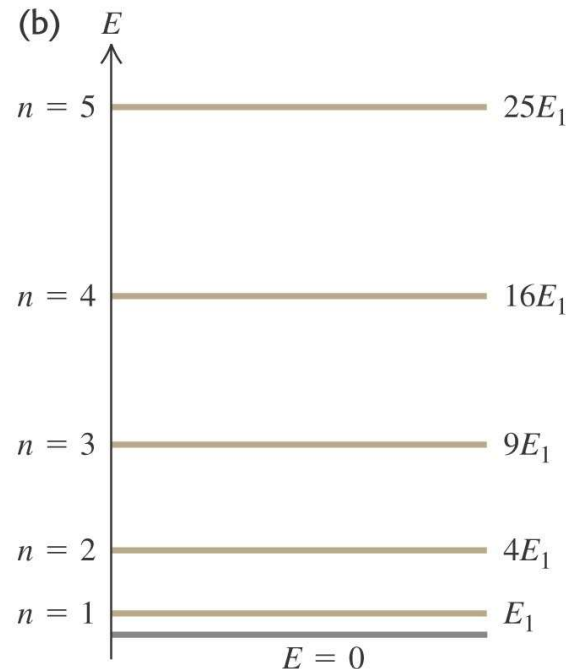
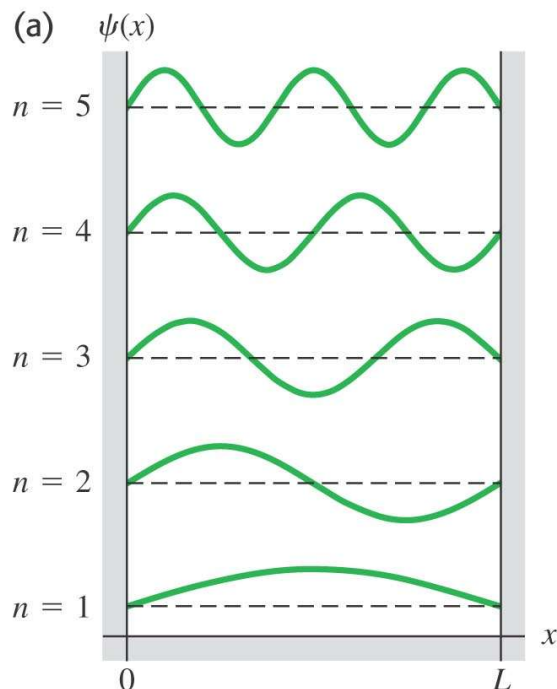


Quantized Energies for a Particle in a Box

Since the wave number k_n is quantized, the energy for the particle in the box is also quantized:

$$E_n = \frac{\hbar^2 k_n^2}{2m} = \frac{\hbar^2 \left(\frac{n\pi}{L}\right)^2}{2m} = \frac{n^2 \pi^2 \hbar^2}{2mL^2} \left(\text{or } = \frac{n^2 h^2}{8mL^2}\right), \quad n = 1, 2, 3, \dots$$

(n is called the **quantum number**)



Note: the lowest energy with $n = 1$ is not zero:

$$E_1 = \frac{h^2}{8mL^2} > 0$$

$n = 0$ gives $\psi(x) = 0$ and it means *no* particle.